organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Li-Rong Wen,^a* Zi-Qin Ke,^b Ming Li^a and Wei-Si Guo^a

^aCollege of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, Qingdao 266042, Peoples' Republic of China, and ^bDepartment of Chemistry, Xingjiang Normal University, Urumuqi 830000, Peoples' Republic of China

Correspondence e-mail: wenlirong@126.com

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å Disorder in main residue R factor = 0.037 wR factor = 0.100 Data-to-parameter ratio = 11.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound, $C_{13}H_{13}N_5S$, possesses crystallographically imposed mirror symmetry. In the crystal structure, the molecules interact through strong intermolecular $N-H\cdots N$ hydrogen bonds to form ribbons running parallel to the *a* axis.

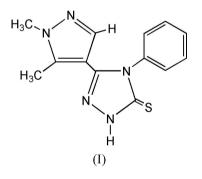
3-(1,5-Dimethylpyrazol-4-yl)-4-phenyl-

1H-1,2,4-triazole-5(4H)-thione

Received 29 March 2005 Accepted 15 April 2005 Online 23 April 2005

Comment

In recent years, 1,2,4-triazole derivatives have been extensively studied due to their broad spectrum of biological activities, the most interesting of which are microbicidal (Ismalel *et al.*, 1984), antifungal (Mishra *et al.*, 1991), antibacterial (Mohan *et al.*, 1990), anti-inflammatory (Rani *et al.*, 1990) and plant-growth regulation activities (Wang *et al.*, 1998). Pyrazoles have been widely investigated due to their close association with various biological effects (Hu *et al.*, 2002). In the course of our systematic studies aimed at the synthesis of new bioactive compounds incorporating these groups, we have synthesized the title compound, (I), the structure of which is reported here.



Bond distances and angles (Table 1) are as expected for this type of compound. The molecule has crystallographically imposed mirror symmetry, with the triazole and pyrazole rings lying on the mirror plane and the phenyl ring strictly orthogonal to it.

In the crystal structure, the molecules are linked through strong $N-H\cdots N$ hydrogen-bond interactions (Table 2) to form ribbons running parallel to the *a* axis (Fig. 2).

Experimental

A mixture of 1,5-dimethyl-4-pyrazolylcarbohydrazide (0.925 g, 6 mmol) and phenyl isothiocynate (0.811 g, 6 mmol) in ethanol (10 ml) was refluxed for 3 h. The solid was separated by filtering after cooling, washed with cold ethanol, dried and recrystallized from ethanol to give N-(1',5'-dimethyl-4'-pyrazolylcarbonyl)-N'-phenyl-3-thiosemicarbazide (0.581 g, 2 mmol), which was refluxed in aqueous sodium carbonate solution (30 ml) for 7 h. The reaction mixture was cooled and acidified to pH 5 with 2 M hydrochloric acid. The resulting

 $\ensuremath{\mathbb{C}}$ 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

1.340 (4)

1.308 (3)

1.450 (4)

1.381 (4)

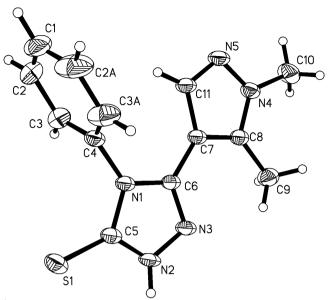


Figure 1

View of the title compound, with 35% probability displacement ellipsoids. Only one disorder component is shown. [Symmetry code (A): $x, \frac{1}{2} - y, z$.]

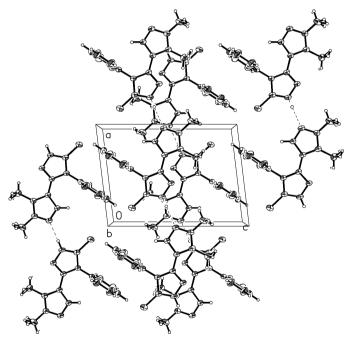


Figure 2

The molecular packing of the title compound, viewed along the b axis. Hydrogen bonds are shown as dashed lines.

white crude precipitate was filtered, washed with cold water, dried, and recrystallized from DMF to give the title compound (m.p. 573 K).

Crystal data

$C_{13}H_{13}N_5S$	$D_x = 1.365 \text{ Mg m}^{-3}$
$M_r = 271.35$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/m$	Cell parameters from 912
a = 8.363 (7) Å	reflections
b = 6.898(6) Å	$\theta = 2.5 - 23.0^{\circ}$
c = 11.524 (10) Å	$\mu = 0.24 \text{ mm}^{-1}$
$\beta = 96.879 \ (12)^{\circ}$	T = 293 (2) K
$V = 660.0 (10) \text{ Å}^3$	Block, colourless
Z = 2	$0.32 \times 0.24 \times 0.16 \text{ mm}$

Data collection

Bruker APEXII CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1996) $T_{min} = 0.933, T_{max} = 0.962$ 3602 measured reflections	1266 independent reflections 972 reflections with $I > 2\sigma(I)$ $R_{int} = 0.024$ $\theta_{max} = 25.0^{\circ}$ $h = -8 \rightarrow 9$ $k = -8 \rightarrow 8$ $l = -13 \rightarrow 13$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.100$ S = 1.07 1266 reflections 111 parameters	$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.061P)^2 \\ &+ 0.199P] \\ &where \ P = (F_o^2 + 2F_c^2)/3 \\ &(\Delta/\sigma)_{\max} < 0.001 \\ &\Delta\rho_{\max} = 0.18 \ e^{\ \text{\AA}^{-3}} \\ &\Delta\rho_{\min} = -0.19 \ e^{\ \text{\AA}^{-3}} \end{split}$

Table 1 Selected geometric parameters (Å, °).

H-atom parameters constrained

S1-C5 1.662 (3) N4-C8 1.373 (3) N5-C11 N1-C5 N1-C61.374 (3) C6 - C7N2 - C51.319 (3) C7-C8

N3-C6	1.298(3) $C/-C11$		1.399 (3)	
05 NH 06	100.0 (2)	NO 05 01	107.00 (10)	
C5-N1-C6	108.0 (2)	N2 - C5 - S1	127.38 (19)	
C5-N2-N3	114.5 (2)	N1-C5-S1	129.5 (2)	
C6-N3-N2	103.9 (2)	N3-C6-N1	110.5 (2)	
C8-N4-N5	112.3 (2)	C8-C7-C11	104.6 (2)	
C11-N5-N4	105.5 (2)	N4-C8-C7	106.3 (2)	
N2-C5-N1	103.1 (2)	N5-C11-C7	111.4 (3)	

Table 2		
Hydrogen-bonding geometry	(Å,	°).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N2-H2A\cdots N5^{i}$	0.86	1.94	2.803 (2)	179.5
Symmetry code: (i) 1	$\perp r v \tau$			

Sy hetry code: (i) 1 + x, y, z

All H atoms were placed in calculated positions, with C-H = 0.93-0.96 Å, N-H = 0.91 Å, and refined using a riding model, with $U_{iso}(H)$ = 1.2 or 1.5 (for methyl H atoms) times $U_{eq}(C,N)$. The H atoms on C9 and C10 were found to be disordered. They were refined over two positions (occupancies of 0.5 for each atom).

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

This project was supported by the Natural Science Foundation of Shandong Province (No. Y2003B01).

References

Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.

- Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Hu, G. Q., Huang, G. Q., Zhang, H. B. & Wang, H. (2002). Chin. J. Org. Chem. **22**, 667–671.

- Ismalel, A. M., Yousif, M. Y., Metwally, M. A. & El-kerdawy, M. M. (1984). Indian Chem. Soc. B, 23, 489–491.
- Mishra, R. K., Tewari, R. K., Shivastava, S. K. & Bahel, S. C. (1991). J. Indian Chem. Soc. 68, 110–112.
- Mohan, J., Anjaneyulu, G. S. R., Verma, P. & Yamini, K. V. S. (1990). J. Indian Chem. Soc. B, 29, 89–90.
- Rani, B. R., Bhalerao, U. T. & Rahman, M. F. (1990). J. Indian Chem. Soc. B, 29, 995–998.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wang, Y. G., Cao, L., Yan, J., Ye, H. F. & Zhou, Q. C. (1998). Chem. J. Chin. Univ. 19, 1247–1250.